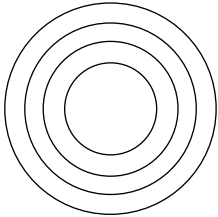
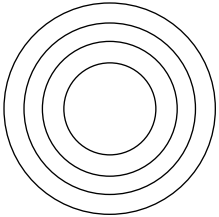
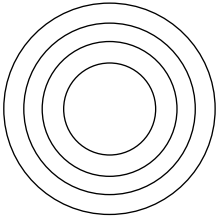
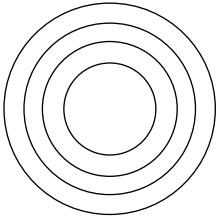
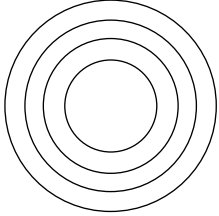
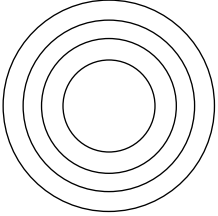


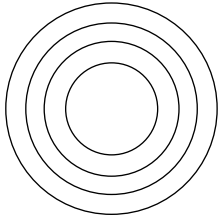
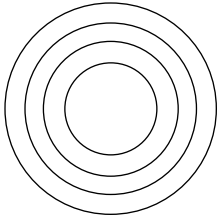
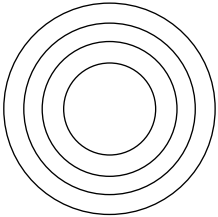
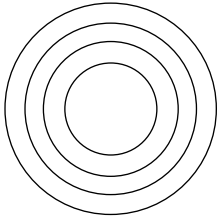
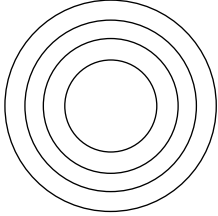
Bohr Model Practice

Use the information provided for each element to draw Bohr Model diagrams. Rather than drawing individual protons and neutrons, you may simply label how many of each there are in the nucleus (e.g. He: 2p, 2n). Then, draw the individual electrons on the appropriate energy levels (keep in mind the maximum number of electrons allowed on each level).

1. Beryllium : atomic # 4, mass 9 	2. Sodium : atomic # 11, mass 23 
3. Sulfur – atomic #: 16, mass 32 	4. Fluorine – atomic #: 9, mass 19 
5. Calcium – atomic #: 20, mass 40 	6. Argon – Atomic #: 18, mass 42 

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