

Write the formulas of the following *covalent* compounds:

- |     |                          |  |
|-----|--------------------------|--|
| 31) | nitrogen trichloride     | <b><math>\text{NCl}_3</math></b>         |
| 32) | boron carbide            | <b><math>\text{BC}</math></b>            |
| 33) | dinitrogen trioxide      | <b><math>\text{N}_2\text{O}_3</math></b> |
| 34) | phosphorus pentafluoride | <b><math>\text{PF}_5</math></b>          |
| 35) | methane                  | <b><math>\text{CH}_4</math></b>          |
| 36) | sulfur dibromide         | <b><math>\text{SBr}_2</math></b>         |
| 37) | diboron tetrahydride     | <b><math>\text{B}_2\text{H}_4</math></b> |
| 38) | oxygen difluoride        | <b><math>\text{OF}_2</math></b>          |
| 39) | carbon disulfide         | <b><math>\text{CS}_2</math></b>          |
| 40) | nitrogen monoxide        | <b><math>\text{NO}</math></b>            |

Write the formulas for the following *covalent* compounds:

- 1) antimony tribromide \_\_\_\_\_
- 2) hexaboron silicide \_\_\_\_\_
- 3) chlorine dioxide \_\_\_\_\_
- 4) hydrogen iodide \_\_\_\_\_
- 5) iodine pentafluoride \_\_\_\_\_
- 6) dinitrogen trioxide \_\_\_\_\_
- 7) ammonia \_\_\_\_\_
- 8) phosphorus triiodide \_\_\_\_\_